

# Dielectric ceramic and nanopowders of holmium-doped BaTiO<sub>3</sub> prepared by sol-gel combustion

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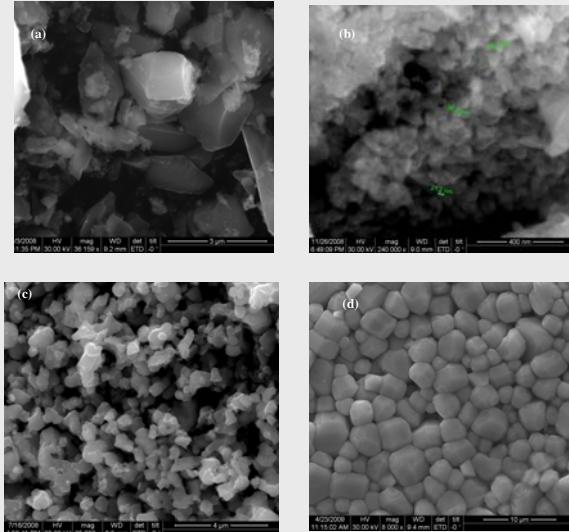
**Aim:** Investigation of the structure and dielectric properties of derived Ba<sub>0.97</sub>Ho<sub>0.03</sub>TiO<sub>3</sub> ceramics prepared by sol-gel combustion

## Sol-gel synthesis of Ba<sub>0.97</sub>Ho<sub>0.03</sub>TiO<sub>3</sub>

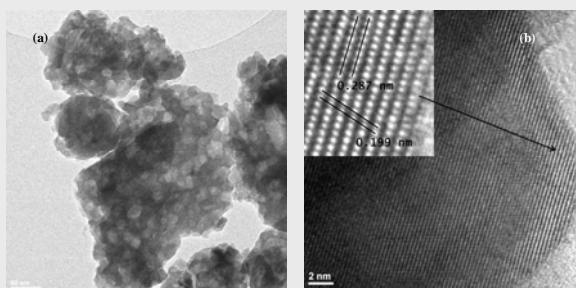
### Synthesis conditions

- Ba precursor: barium acetate Ba(CH<sub>3</sub>CO<sub>2</sub>)<sub>2</sub>
- Ti precursor: titanium (IV) isopropoxide Ti(OC<sub>3</sub>H<sub>7</sub>)<sub>4</sub>
- Ho precursor: holmium oxide Ho<sub>2</sub>O<sub>3</sub>
- Sol aq. 1M Ba(CH<sub>3</sub>CO<sub>2</sub>)<sub>2</sub>
- Holmium oxide dissolved in H<sub>2</sub>O+HNO<sub>3</sub>
- Peroxo-citrate complex of titanium isopropoxide:
  - titanium (IV) isopropoxide;
  - nitric acid;
  - citric acid;
  - hydrogen peroxide;
  - water.
- Molar ratio citric acid:metal cations = 2.5:1
- Molar ratio citrate:nitrate (CA/NO<sub>3</sub>)=1.3.
- pH: 5; (ammonium hydroxide).
- Refluxing temperature: ~75°C
- Dry gel temperature: 100 °C
- Firing temperature of the gel to obtain Ba<sub>0.97</sub>Ho<sub>0.03</sub>TiO<sub>3</sub>: 1100 °C
- Sintering temperature to obtain BT-Ho<sub>0.03</sub> ceramic: 1350 °C, 2h.

## Microstructures analysis

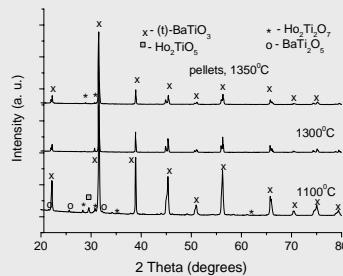


**Fig.1.** SEM images of BT-Ho<sub>0.03</sub> precursor gel dried at 100 °C (a), fired at 700 °C (b) and 1100 °C (c) and, ceramic sintered at 1350 °C, 2h in air (d). Average grains size: 30 nm (b), 250 nm (c) and 2 μm (d)



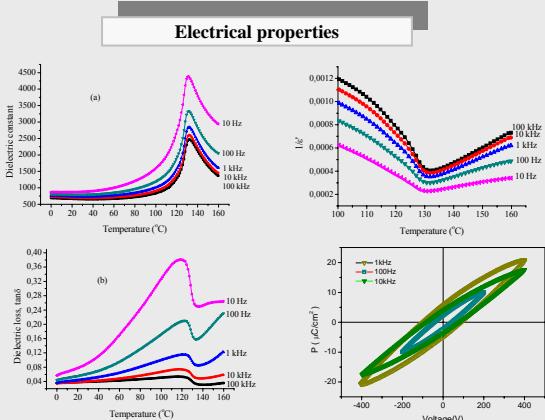
**Fig.2.** TEM and HR-TEM micrographs of the (Ba, Ho)TiO<sub>3</sub> gel heated at 700 °C

## Structure analysis



### X-ray diffraction analysis results

- Ba<sub>0.97</sub>Ho<sub>0.03</sub>TiO<sub>3</sub> cryst. on the t-BaTiO<sub>3</sub> lattice, BaTi<sub>2</sub>O<sub>5</sub>, Ho<sub>2</sub>TiO<sub>5</sub> and Ho<sub>2</sub>Ti<sub>2</sub>O<sub>5</sub>, at 700-1100 °C (powder);
- Ba<sub>0.97</sub>Ho<sub>0.03</sub>TiO<sub>3</sub> cryst. on the t-BaTiO<sub>3</sub> lattice, and Ho<sub>2</sub>Ti<sub>2</sub>O<sub>5</sub>, at 1300 °C (powder);
- Ba<sub>0.97</sub>Ho<sub>0.03</sub>TiO<sub>3</sub> cryst. on the t-BaTiO<sub>3</sub> lattice, and Ho<sub>2</sub>Ti<sub>2</sub>O<sub>5</sub>, at 1350 °C (ceramic).



**Fig.4.** Variation of dielectric constant and dielectric loss of BT-Ho<sub>0.03</sub> ceramics with temperature and frequency

## Conclusions

- Powder of BaTiO<sub>3</sub> doped with 3 mol% Ho, consisting of Ba<sub>0.97</sub>Ho<sub>0.03</sub>TiO<sub>3</sub> as main phase was obtained by heating the precursor gel at 700 °C.
- The as-prepared powder is composed of particles with agglomerated structures; the average size of the crystallites was 30 nm.
- The XRD patterns of the powders and sintered BT-BT<sub>0.03</sub> ceramics indicated a material composed of Ba<sub>0.97</sub>Ho<sub>0.03</sub>TiO<sub>3</sub> crystallized on the tetragonal-BaTiO<sub>3</sub> lattice, and traces of Ho<sub>2</sub>Ti<sub>2</sub>O<sub>5</sub>.
- The BT-Ho<sub>0.03</sub> ceramics prepared by sol-gel auto-combustion, presented good dielectric properties.

## References

- [1] M. Cernea, G. Montanari, C. Galassi and A. Costa: Synthesis of La and Nb doped PZT powder by the gel-combustion method, *Nanotechnology*, 17, 1731 (2006).
- [2] M. Cernea, E. Andronescu, R. Radu, F. Fochi, C. Galassi, "Sol-gel synthesis and characterization of BaTiO<sub>3</sub> doped-(Bi<sub>1/2</sub>Na<sub>1/2</sub>)TiO<sub>3</sub> piezoelectric ceramics", *Journal of Alloys and Compound Materials*, 490 (2010) 690-694.

**Table 2.** T<sub>c</sub> and T<sub>0</sub> obtained from the slope of 1/ε' vs T plot at various frequencies for Ba<sub>0.97</sub>Ho<sub>0.03</sub>TiO<sub>3</sub> ceramic

Frequency	10 Hz	100 Hz	1 kHz	10 kHz	100 kHz
T <sub>c</sub> =303.94 °C	C <sub>R1</sub> =62 pF		C <sub>R1</sub> =52 pF		
T <sub>c</sub> =273.15 °C	C <sub>1</sub> = 63 pF	TCC <sub>1</sub> =5.38·10 <sup>-4</sup>	C <sub>1</sub> = 56 pF	TCC <sub>1</sub> =2.56·10 <sup>-3</sup>	
T <sub>c</sub> =333.76 °C	C <sub>2</sub> =68 pF	TCC <sub>2</sub> =-3.31·10 <sup>-3</sup>	C <sub>2</sub> =53 pF	TCC <sub>2</sub> =-6.41·10 <sup>-4</sup>	
		TCC=-1.33·10 <sup>-3</sup>		TCC=0.96·10 <sup>-3</sup>	