

# Photocatalytic degradation of Eosin Y dye using SnO<sub>2</sub> nanocomposites

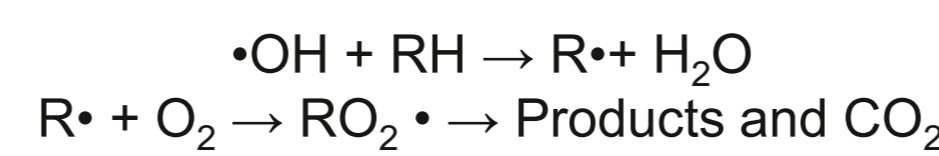
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The tin oxide based nanocomposites have been used as efficient and environmentally benign catalysts. The developed protocols using this kind of material are advantageous in terms of simple experimentation, reusable catalyst, excellent yields of the products, short reaction time and preclusion of toxic solvents. In this work, we have reported novel synthesis and characterization of supported SnO<sub>2</sub> catalysts.

## INTRODUCTION

The elimination of toxic chemicals from wastewaters is presently one of the most important subjects in pollution control. An alternative to conventional water treatment processes is the advanced oxidation processes using solar radiation. [1-4]. Advanced oxidation processes involve the generation of hydroxyl ( $\bullet\text{OH}$ ) radicals which oxidize the pollutants. After fluorine, the hydroxyl radical is the second strongest known oxidant having an oxidation potential of 2.8 eV. It is able to oxidize and mineralize almost every organic molecule, yielding CO<sub>2</sub> and inorganic ions as shown in the following equations:



It is known that tin dioxide is a versatile material, applicable in many physical-chemical processes, being one of the most intensively studied semiconductors. SnO<sub>2</sub> is an n-type semiconductor with a direct band gap of 3.6 eV between the full oxygen 2p valence band and the tin states at the bottom of the conduction band.

## EXPERIMENTAL

### Preparation of SnO<sub>2</sub> nanometer powders

Bentonite montmorillonitic-clay, provided by firma Riedel-de Haen Chemicals Company, was used as starting material. The nanosize SnO<sub>2</sub> particles were prepared by two methods:

- Impregnation method;
- Pillaring method (dispersing the tin oxide particles).

The novel materials were successfully synthesized in ethanol/water solution. By changing the reaction conditions, the size and the morphology can be controlled. In summary, we find an economical and efficient process for synthesizing SnO<sub>2</sub> nanocomposites. The concentration of NaOH and the ratio of water and ethanol are important to synthesize various morphologies of SnO<sub>2</sub> crystals.

These samples were investigated the photocatalytic oxidative degradation and discoloration of synthetic wastewater containing Eosin Y dye, varying the concentration of catalyst (0,5-2g/l).

### Studies of photocatalysis

By changing the reaction conditions, the size and the morphology can be controlled. In summary, we find an economical and efficient process for synthesizing SnO<sub>2</sub> montmorillonite nanocomposites. The concentration of NaOH and the ratio of water and ethanol are important to synthesize various morphologies of SnO<sub>2</sub> crystals. These samples were investigated the photocatalytic oxidative degradation and discoloration of synthetic wastewater containing Eosin Y dye, varying the concentration of catalyst (0,5-2g/l).

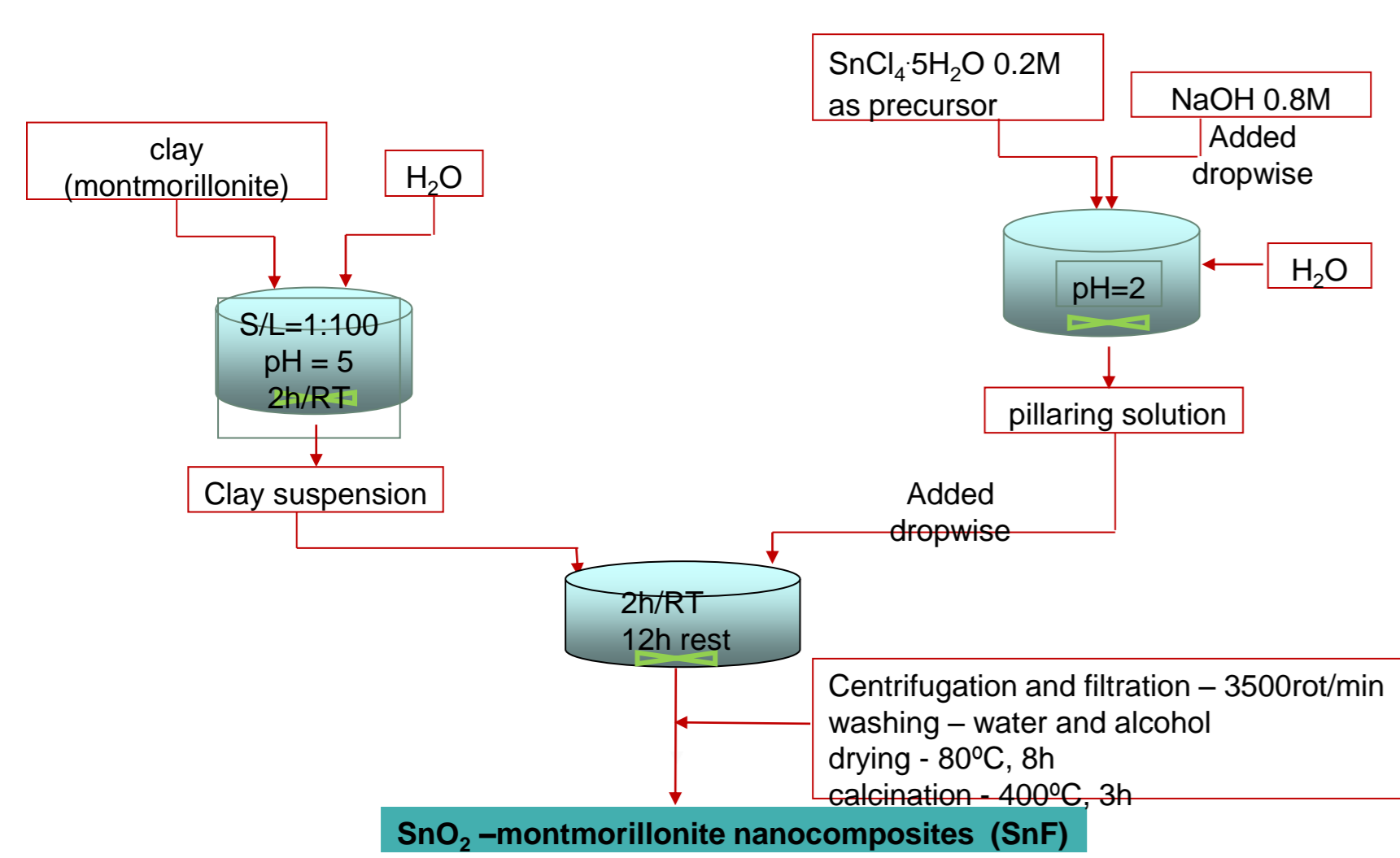


Figure 1: Preparation of SnO<sub>2</sub> nanometer powders by pillaring method

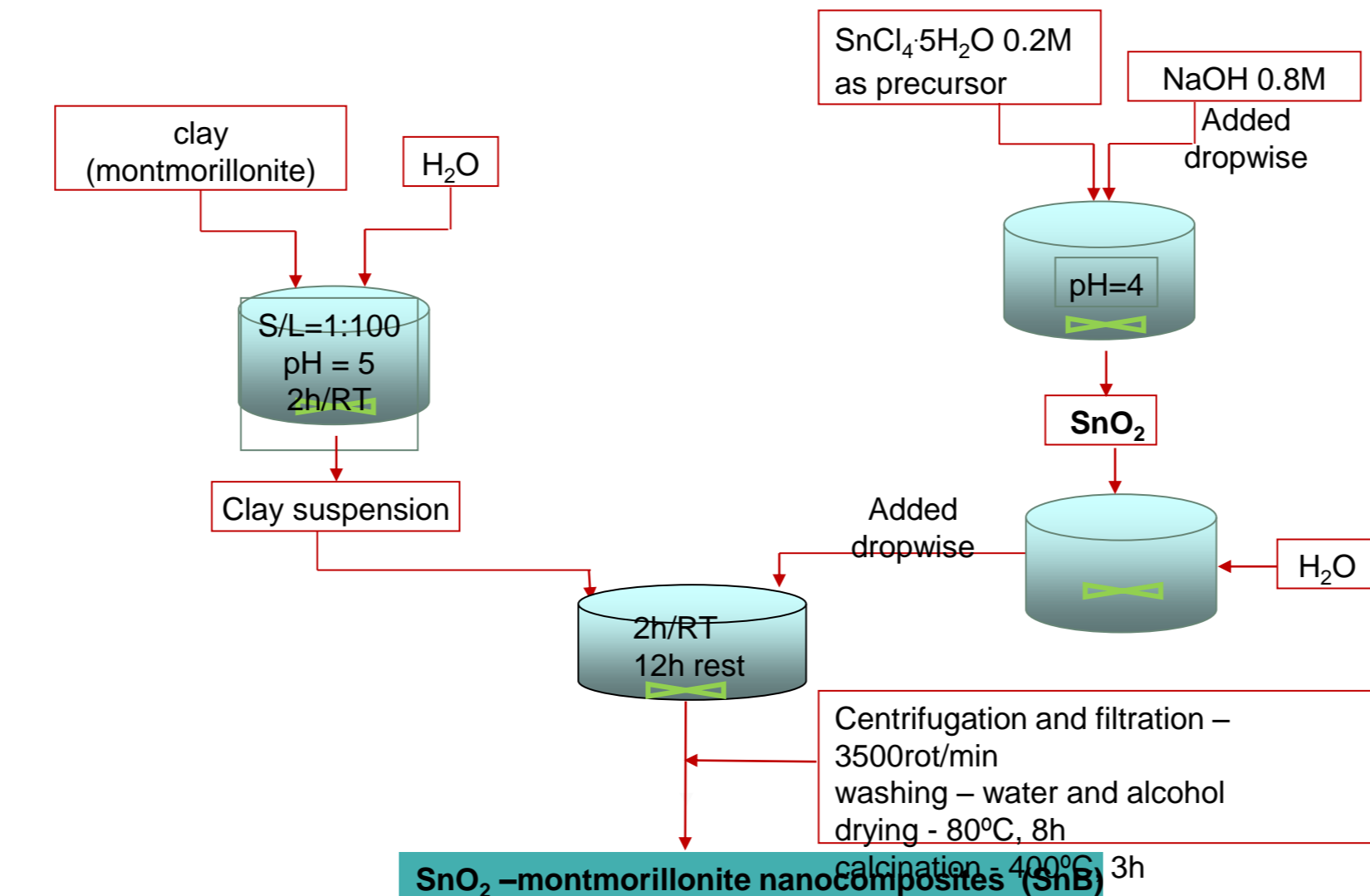


Figure 2: Preparation of SnO<sub>2</sub> nanometer powders by impregnation method

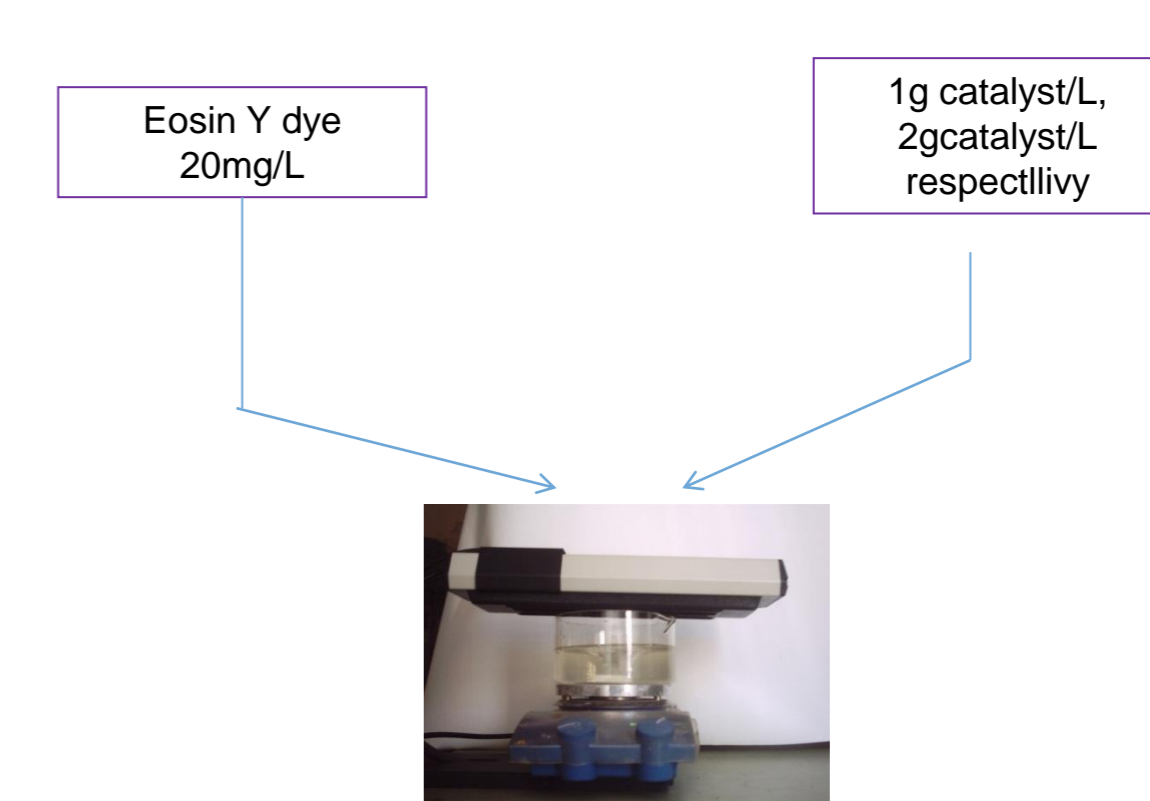


Figure 3: Studies of photocatalysis

Photocatalysis occurs in two stages:  
- adsorption in the dark for 30 minutes;  
- measured dye concentration, which is exposed to UV light wavelength of 254 nm at equal time intervals based on the calibration curve.

## Characterization methods

The structure and properties of the obtained materials were studied by X-ray diffraction (XRD), FTIR spectroscopy, N<sub>2</sub> adsorption-desorption isotherms and UV-Vis diffuse reflectance spectroscopy. The structures of pure and tin-oxide pillared clay were investigated using Shimadzu LabX XRD 6000 diffractometer. The diffraction angle was scanned from 10 to 80 degrees, a usual interval for complex clays and SnO<sub>2</sub>. Measurement Condition was: X-ray tube, Cu target Cu, voltage = 40.0 (kV), current 30.0 (mA) and scanning: scan mode-Continuous Scan, scan speed 2.0000 (deg/min). FTIR spectra were recorded on a FTIR JASCO 660+ spectrometer. UV-vis diffuse reflectance spectra were studied by Shimadzu UV-2401 PC Recording Spectrophotometer of samples were obtained in the range 200–600 nm. The adsorption isotherms of N<sub>2</sub>, specific surface areas and porosities were determined with a Nova 2200e (Quantachrome Instruments) automated sorptometer at 77 K.

## RESULT AND DISCUSSIONS

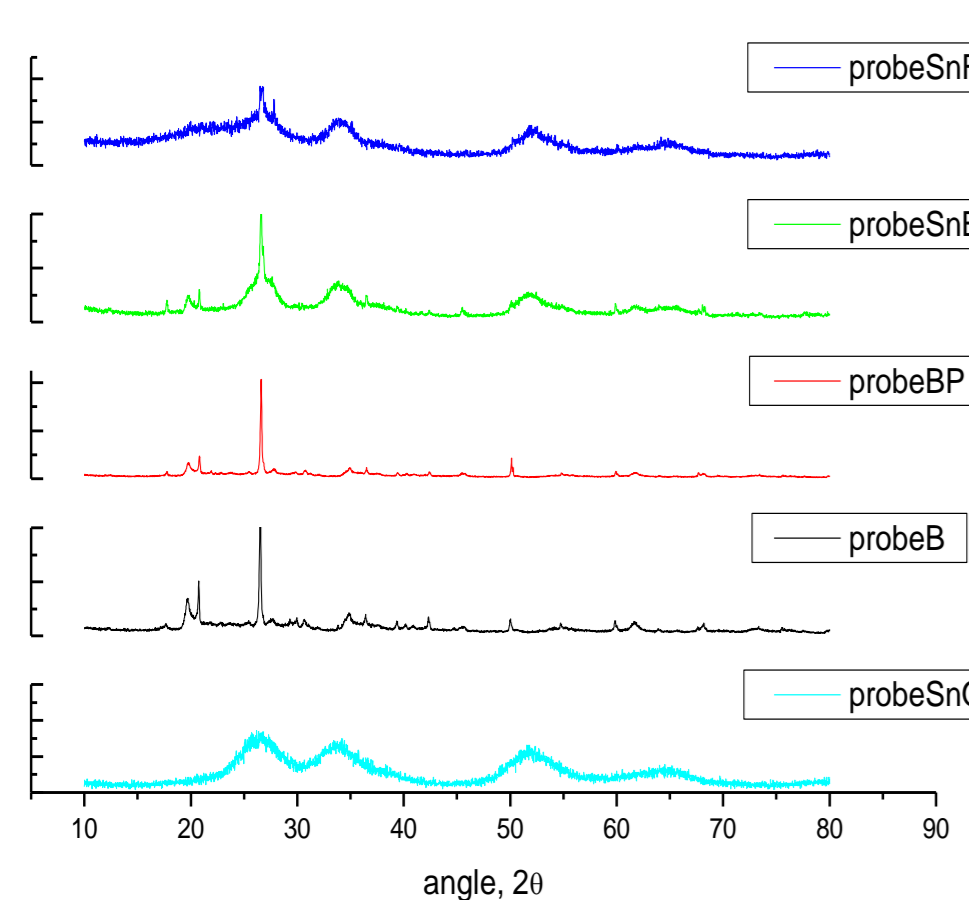


Figure 4: Comparative study between samples X-ray diffraction (XRD) patterns of the studied samples

Identification of peaks was performed using the diffractometer software. The pure clay presents all specific peaks for complex clays: quartz, bentonite, feldspar, aluminum silicate and mica. Reference diffractogram for thin oxide show a nanocrystalline powder with tetragonal structure of elementary cell, having lattice parameters: a=4.7200 Å, b=4.7200 Å, c=3.1700 Å. For this elementary cell dimensions, is well fitted in SiO<sub>2</sub> sites of clay, as identification image show.

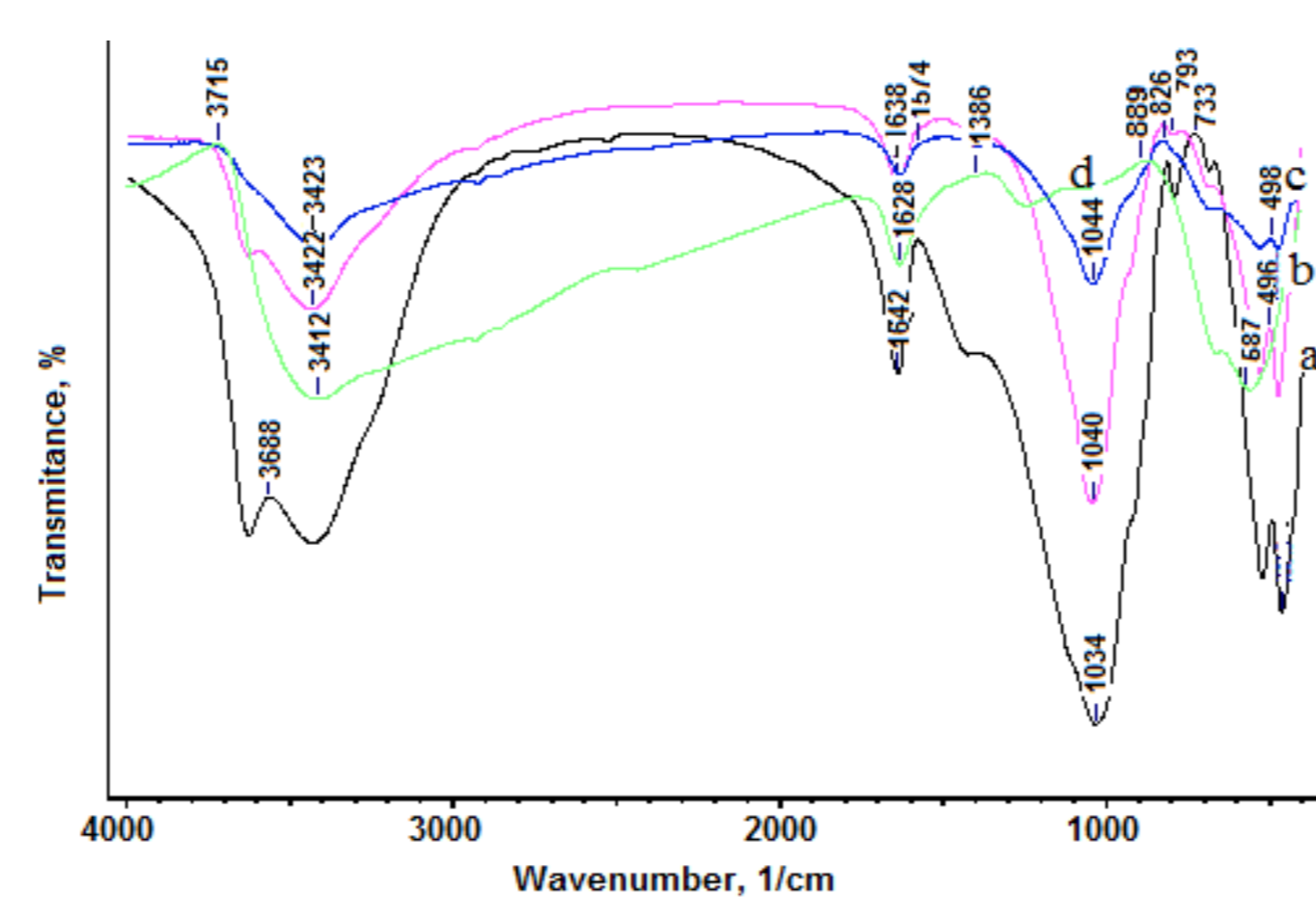


Figure 5: FTIR Spectra: a – clay; b - SnO<sub>2</sub> – clay nanocomposites (SnB); c - SnO<sub>2</sub> – pillared clay nanocomposites (SnF); d- SnO<sub>2</sub> nanoparticles

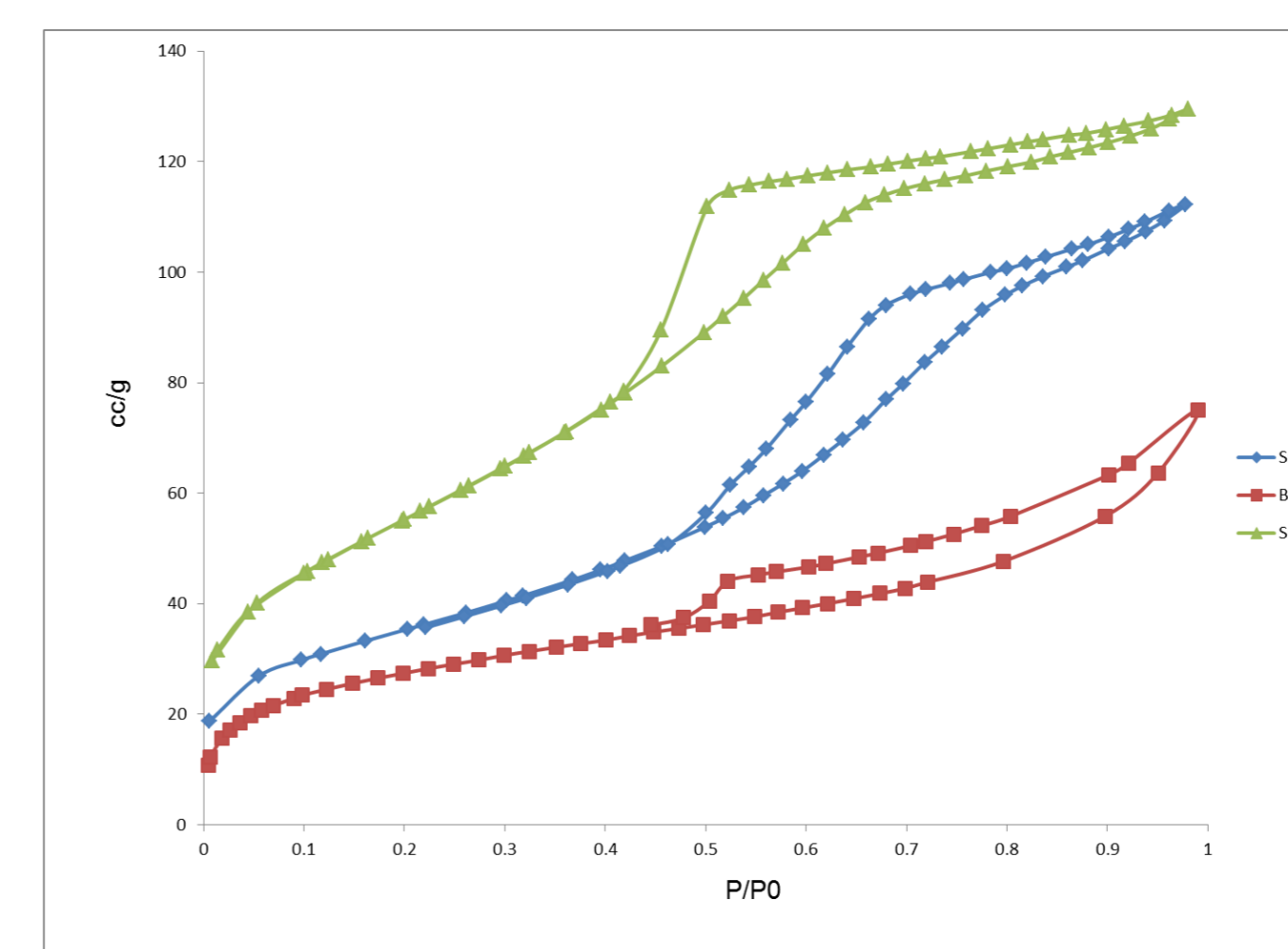


Figure 6: Nitrogen adsorption isotherm at 77k, for studied sample

Sample	Surface area, S <sub>BET</sub> (m <sup>2</sup> /g)	Pore volum e, cc/g	Pore diameter, nm
clay (B)	132.7	0.135	1.038
SnO <sub>2</sub> – clay nanocomposites (SnB)	234.2	0.203	1.155
SnO <sub>2</sub> – pillared clay nanocomposites (SnF)	191.4	0.165	3.851

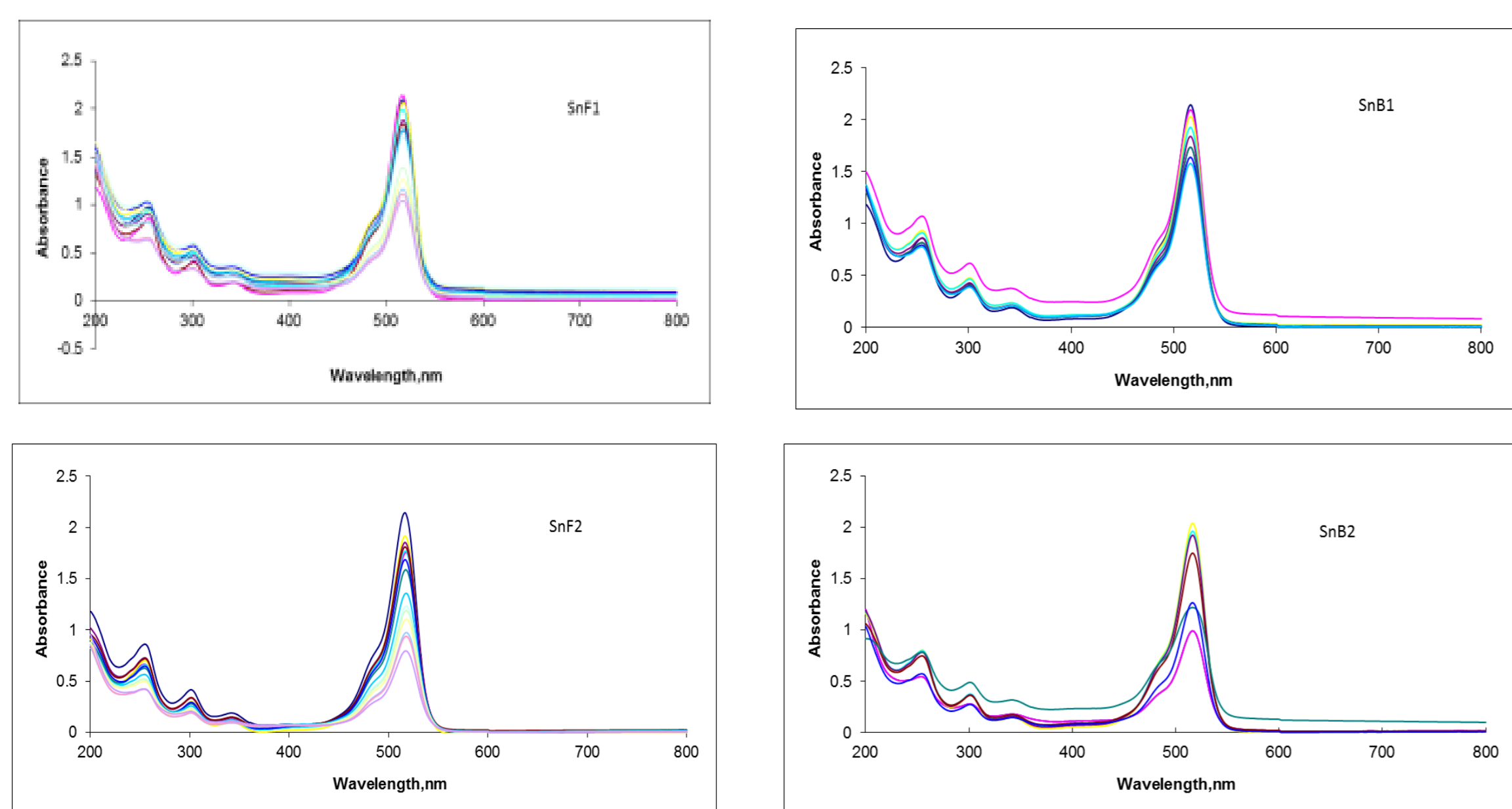


Figure 7: UV-vis spectra of the studied samples

Degradation of Eosin Y dye (EY) – 20mg/L  
electrical double layer of the solid-liquid interface  
adsorption/desorption processes  
separation of the e<sup>-</sup>/h<sup>+</sup> pairs  
EY ⇌ xanthene fluorescent dye

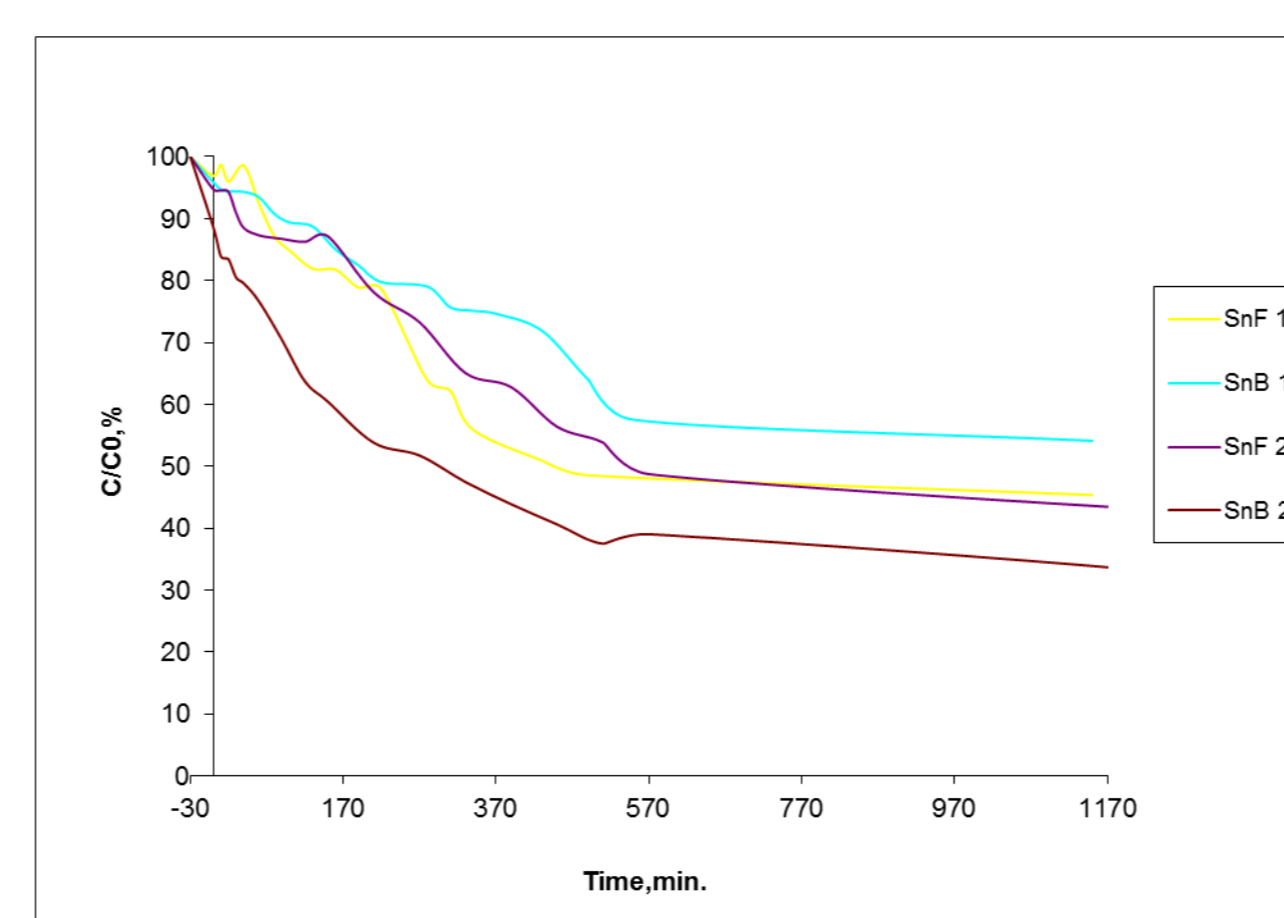


Figure 8: Relative residual concentration of Eosin Y dye (C/C<sub>0</sub>) after treatment with an aqueous solution of SnO<sub>2</sub> clay nanocomposites samples

Results obtained by spectrophotometric method show a decrease (60-70%) in dye concentration over time.

## CONCLUSIONS

- We reported the synthesis and characterization of some new nanocomposites obtained by Impregnation method and pillaring method (dispersing the tin oxide particles) - SnO<sub>2</sub> clay nanocomposites.
- The physico-chemical properties of the studied nanocomposites were characterized by X-ray diffraction (XRD), N<sub>2</sub> adsorption-desorption isotherms, FTIR spectroscopy and UV-Vis diffuse reflectance spectroscopy
- The ability of the nanocomposites to degrade Eosin Y dye under UV light.
- In this work, the photocatalytic oxidative degradation of Eosin Y dye, xanthene fluorescent dye, has been studied under solar radiation. It is observed that nanosized compound SnO<sub>2</sub>/clay photocatalysts are efficient photocatalysts, both in respect of decolorization as well as mineral-ization.
- Nanosized compound SnO<sub>2</sub>/clay photocatalysts synthesized are for use in treatment of organic wastewaters by converting the carcinogenic compounds to harmless compounds.
- Nanoporous materials have been a core focus in nanosciences, which is an ever-growing multidisciplinary field of study attracting tremendous interest, investments and effort in research development around the world

### References

- [1] Muruganandham, M., Swaminathan, M., Solar driven decolorisation of Reactive Yellow 14 by advanced oxidation processes in heterogeneous and homogeneous media. *Dyes and Pigments* 72 (2007) 137-143.
- [2] Duran, A., Monteagudo, I.M., Solar photocatalytic degradation of Reactive Blue 4 using a Fresnel lens. *Water Research* 41 (2007) 690 - 698.
- [3] Wojnarovits, L., Takacs, E., Irradiation treatment of azo dye containing wastewater: An overview. *Radiation Physics and Chemistry* 77(3) (2008) 225-244.
- [4] Ollis, D.F., Pelizzetti, E., Serpone, N., *Photocatalysis: Fundamentals and applications*, Wiley: New York, 1989.

### Acknowledgments

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